







SDS no. CZ2E7RES • Version 1.0 • Date of issue: 2023-03-01

#### **SECTION 1: Identification**

#### **GHS Product identifier**

Product name TIN METAL (Foil, Granules)

#### Recommended use of the chemical and restrictions on use

Tin plated steel containers for food preservation, tin alloys (brasses and bronzes, bell metal, Babbitt metal, die casting alloy, pewter, phosphor bronze, type, and White metal), soldering alloys for electrical/electronic and general industrial applications, specialized alloys such as dental amalgams, titanium alloys used in aircraft engineering, niobium-tin and indium-tin alloys used in superconducting cables and magnets and indium-tin oxide for metallic photonic crystals, low-melting alloys for fire control, organ pipes, tin alloys are important in the production of coatings by electroplating and hot tinning (the most important of these are tin-zinc, tin-nickel, tin-cobalt, and tin-copper), tin coatings (applied to most metal surfaces by electrodeposition, while in hot-dipping, molten tin wets and adheres readily to clean iron, steel, copper, and copper-base alloys), corrosion-resistant coatings (for lead or zinc and steel), cladding, tinned wire (all copper wire that is to be rubber covered), collapsible tubes, anodes for electron plating, manufacture of chemicals (tin salts), block tin (used to coat copper cooking utensils and lead sheet, or to line lead pipe for distilled water, beer, carbonated beverages, and some chemicals), powder metallurgy applications, exothermic welding, catalysts, colours, stabilizer, cast and wrought forms, window glass manufacture (Pilkington process) and transportation applications.

# Supplier's details

Name ChemSupply Australia Pty Ltd

Address 38-50 Bedford Street

5013 Gillman South Australia

Australia

Telephone 08 8440 2000

email www.chemsupply.com

**Emergency phone number** 

CHEMCALL 1800 127 406 (Australia) / +64-4-917-9888 (International)

## **SECTION 2: Hazard identification**

## **General hazard statement**

Not classified as dangerous goods according to the Australian Dangerous Goods Code (ADG).

Classified as non-Hazardous according to the Globally Harmonised System of classification and labelling of Chemicals (GHS) including Work, Health and Safety regulations, Australia.

# Classification of the substance or mixture

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#### GHS classification in accordance with: UN GHS revision 7

Not a hazardous substance or mixture.

#### **GHS** label elements, including precautionary statements

Not a hazardous substance or mixture.

#### Other hazards which do not result in classification

Not a hazardous substance or mixture.

# **SECTION 3: Composition/information on ingredients**

#### **Mixtures**

Molecular weight: 118.69

#### Components

Component	Concentration
TIN (CAS no.: 7440-31-5; EC no.: 231-141-8)	100 - 100 % (weight)
CLASSIFICATIONS: No data available. HAZARDS: No data available.	

## **SECTION 4: First-aid measures**

#### **Description of necessary first-aid measures**

General advice First Aid Facilities: Maintain eye wash and normal washroom facilities.

If inhaled If inhaled, remove from contaminated area. Apply artificial respiration if not breathing.

In case of skin contact If skin or hair contact occurs, remove contaminated clothing and flush skin and hair

with running water.

In case of eye contact If in eyes wash out immediately with water.

If swallowed, do NOT induce vomiting.

#### Most important symptoms/effects, acute and delayed

The most important known symptoms and effects are described in the labelling (see section 2.2) and/or in section 11

#### Indication of immediate medical attention and special treatment needed, if necessary

Treat symptomatically based on judgement of doctor and individual reactions of the patient.

# **SECTION 5: Fire-fighting measures**

#### Suitable extinguishing media

Use fire extinguishing media appropriate for surrounding environment. Use water spray, dry chemical, carbon dioxide, or appropriate foam.

# Specific hazards arising from the chemical

Irritating and/or highly toxic fumes and gases, tin/tin oxides.

# Special protective actions for fire-fighters

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Fire fighters should wear full protective clothing and self-contained breathing apparatus (SCBA) operated in positive pressure mode. Fight fire from safe location.

# **SECTION 6: Accidental release measures**

## Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing vapours, mist or gas. For personal protection see section 8.

#### Methods and materials for containment and cleaning up

Sweep up and shovel. Keep in suitable, closed containers for disposal.

# **SECTION 7: Handling and storage**

#### **Precautions for safe handling**

Avoid ingestion and inhalation of dust/granules/foil. Avoid contact with eyes, skin, and clothing. Avoid prolonged or repeated exposure. Minimize dust generation and accumulation. Keep containers closed when not in use. Use in designated areas with adequate ventilation. Keep away from incompatibles such as oxidizing agents, acids, alkalis.

#### Conditions for safe storage, including any incompatibilities

Store in tightly closed containers, in a cool, dry, well-ventilated area away from incompatible substances. Separated from oxidising agents.

# **SECTION 8: Exposure controls/personal protection**

#### **Control parameters**

#### CAS: 7440-31-5

TIN

 $AU/SWA\ (Australia): 2\ mg/m3\ TWA\ inhalation;\ NIOSH: 2\ mg/m3;\ except\ tin\ oxides\ REL\ inhalation;\ 0.1\ mg/m3Fxcept\ Cyhexatin\ REL\ inhalation;\ Oxides\ REL\ inhala$ 

#### Appropriate engineering controls

Use ventilation adequate to keep exposures (airborne levels of dust, fume, vapor, gas, etc.) below recommended exposure limits.f the engineering controls are not sufficient to maintain concentrations of vapours/mists below the exposure standards, suitable respiratory protection must be worn.

#### Individual protection measures, such as personal protective equipment (PPE)

#### **Eye/face protection**

The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate. Must comply with Australian Standards AS 1337 and be selected and used in accordance with AS 1336.

#### Skin protection

Clean impervious clothing should be worn. Clothing for protection against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals.

Hand Protection: Normally not required but if in doubt ensure hand protection should complies with AS 2161, Occupational protective gloves - Selection, use and maintenance.

#### **Body protection**

Suitable protective workwear, e.g. cotton overalls buttoned at neck and wrist is recommended. Chemical resistant apron is recommended where large quantities are handled.

# **Respiratory protection**

If engineering controls are not effective in controlling airborne exposure then an approved respirator with a replaceable vapor/ mist filter should be used. Refer to relevant regulations for further information concerning respiratory protective requirements. Reference should be made to Australian Standards AS/ NZS 1715, Selection, Use and Maintenance of Respiratory Protective Devices; and AS/NZS 1716, Respiratory Protective Devices, in order to make any necessary changes for individual circumstances.

# **SECTION 9: Physical and chemical properties**

#### Basic physical and chemical properties

Physical state Appearance

Color Odor

Odor threshold

Melting point/freezing point

Boiling point or initial boiling point and boiling range

Flammability

Lower and upper explosion limit/flammability limit

Flash point

**Explosive properties** 

Auto-ignition temperature
Decomposition temperature

Oxidizing properties

рΗ

Kinematic viscosity

Solubility

Partition coefficient n-octanol/water (log value)

Vapor pressure Evaporation rate

Density and/or relative density

Relative vapor density

#### **Particle characteristics**

No data available.

## Supplemental information regarding physical hazard classes

Surface Tension: 544 mN/m (@ 231.9 °C).

#### **Further safety characteristics (supplemental)**

Other Information: Tin has two allotropic forms at normal pressure; at -40 °C crumbles to gray amorphous powder (gray tin; alpha); slowly changes back above 20 °C to white tin (beta); brittle @ 200 °C.

Solid

Almost silver-white to gray, lustrous, malleable metal; cubic (gray); tetragonal (white); rhomboidal (tin brittle); granules, foil, or powder. At -40 °C crumbles to gray amorphous powder (gray tin); slowly changes back above 20 °C to white tin; brittle

at 200 °C.

No data available.

Odourless.

No data available.

231.9 °C.

2270 °C: 2507 °C: 2602 °C.

No data available.

Flammable Limits - Lower: > 99.99 % (powder).

No data available.

The finely divided dust may form flammable/explosive mixtures with air. It may present a dust explosion hazard in the presence of an ignition source. Minimum explosible concentration: 0.19 g/l. Particle size and air concentration determine reactivity.; Tin reacts violently or explosively with fused ammonium nitrate below 200 °C.; Contact of metallic tin

with turpentine may cause fires and explosions. Dust Cloud: 630 °C; Dust Layer: 430 °C.

No data available. No data available. No data available.

Viscosity: 1.85 mPa.s (cP) @ 240 °C.

Solubility in Water: Insoluble in water. Solubility in Organic Solvents: Soluble in hydrochloric acid, sulfuric acid, aqua

regia, alkali; slightly soluble in dilute nitric acid.

No data available. 1.3332 hPa at 1492 °C. No data available.

Specific Gravity: 7.265 (white); 5.769 (gray).

No data available.

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Transformation temp: (beta in equilibrium with alpha) 13.2 °C.

Resistivity of white tin: 11.0  $\mu$ -0hm cm @ 0 °C; 15.5  $\mu$ -0hm cm @ 100 °C; 20.0  $\mu$ -0hm cm @ 200 °C; 22.0  $\mu$ -0hm cm @ mp (solid); 45.0  $\mu$ -0hm cm @ mp (liquid).

# **SECTION 10: Stability and reactivity**

#### Reactivity

Stable under normal conditions of storage and handling.

Reacts with incompatible materials

#### **Chemical stability**

Stable against air and water under normal temperatures, pressures and conditions of handling and storage. Powder oxidizes, especially in the presence of air and moisture. At high temperatures, it burns with an intensive white flame and forms tin oxide.

#### Possibility of hazardous reactions

Possibility of hazardous reactions: Reacts with strong oxidants. Tin reacts violently or explosively with fused ammonium nitrate below 200 °C. In the presence of water, cupric nitrate and tin foil, on prolonged and intimate contact, will produce flaming and sparking. Sodium peroxide and potassium peroxide, potassium dioxide, oxidize tin with incandescence. Reacts violently with strong acids and some extinguishing agents such as bicarbonate powder and carbon dioxide. Reactive with alkalis. The violent reaction between tin and bromine is controlled in halocarbon solutions. Tin /begins to burn at 100 °C in fluorine. Reaction with chlorine trifluoride, in the presence of carbon, is violent. Reacts violently with bromine trifluoride. Tin reacts violently with lodine Bromide. When heated in Chlorine, Tin reacts, producing light and much heat. Interaction with carbon tetrachloride, in the presence of water vapour, is violent. Interaction with disulfur dichloride is violent. Reaction with sulfur is vigorous and accompanied by incandescence. Contact of metallic tin with turpentine may cause fires and explosions. The reaction between tin and tellurium attains incandescence. Experiments involving explosions of molten tin and water are described. The finely divided dust may form flammable/explosive mixtures with air. It may present a dust explosion hazard in the presence of an ignition source, when exposed to heat or by spontaneous chemical reaction with Br2, BrF3, S, Cl2,ClF3, Cu(NO3), K2O2.

#### **Conditions to avoid**

Excess heat, flames, ignition sources, dust generation and incompatible materials. Powder is air and moisture sensitive.

#### **Incompatible materials**

Oxidizing agents (sodium peroxide and potassium peroxide, potassium dioxide, fused ammonium nitrate below 200 °C, cupric nitrate, in the presence of water), strong acids (generation of hydrogen) (nitric acid, hydrochloric acid), strong bases, halogens and halogen trifluorides (fluorine at 100 °C, heat + chlorine, tin chloride, carbon tetrachloride, in the presence of water vapour, bromine, chlorine trifluoride in the presence of carbon, disulfur dichloride), sulfur, some extinguishing agents such as bicarbonate powder and carbon dioxide, tellurium, turpentine, water + heat, mixtures with air in the presence of an ignition source.

#### **Hazardous decomposition products**

Toxic and/or irritating fumes and gases, tin/tin oxides.

# **SECTION 11: Toxicological information**

#### Information on toxicological effects

#### **Acute toxicity**

May be harmful if swallowed. Ingested metallic tin exhibits only moderate toxicity due to poor absorption from the digestive tract and rapid tissue turnover. Ingestion of large doses of powdered tin may cause gastrointestinal irritation, nausea, cramps, vomiting, and diarrhoea (which may be from irritant or astringent action on the stomach), but not permanent injury. Inorganic tin salts, which may form with corrosion depending on a number of factors, including the presence of oxidising agents (oxygen, nitrate) and acids, may cause nausea, vomiting and diarrhoea, may interfere with various enzyme systems and may cause systemic effects on the central nervous system, heart and liver, if ingested in concentrations in excess of 300-500 mg/kg.

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Inhalation of tin dust may cause irritation, due to mechanical action, to nose, throat and respiratory tract, with coughing. Inhaled dust or fumes may cause benign, symptomless pneumoconiosis (stannosis). This form of pneumoconiosis produces distinctive progressive x-ray changes of the lung as long as exposure persists, but there is no distinctive fibrosis, no evidence of disability, and no special complicating factors. Inhalation of fumes may cause metal fume fever, which is characterized by flu-like symptoms with metallic taste, fever, chills, cough, weakness, chest pain, muscle pain and increased white blood cell count.

#### Skin corrosion/irritation

May cause mild skin irritation, resulting in redness and itchiness.

# Serious eye damage/irritation

Dust may cause eye irritation due to mechanical action, with redness and pain. Granules and foil may cause eye damage due to mechanical action.

#### Respiratory or skin sensitization

No data available

#### Germ cell mutagenicity

No data available.

#### Carcinogenicity

No data available.

#### Reproductive toxicity

Tin [resp/skin]: animal-possible increase in subtle neurological & skeletal deformities (from: <qt>Reproductive Hazards of the Workplace<qt> by Linda M. Frazier, MD, MPH & Marvin L. Hage, MD).

#### Summary of evaluation of the CMR properties

No data available.

# Specific target organ toxicity (STOT) - single exposure

No data available.

#### Specific target organ toxicity (STOT) - repeated exposure

No data available.

# **Aspiration hazard**

No data available.

#### **Additional information**

Chronic Effects: Prolonged and/or repeated contact may cause irritation and/or dermatitis. Chronic exposure to dust or fumes may have effects on the lungs, resulting in a benign pneumoconiosis (stannosis). This form of pneumoconiosis produces distinctive progressive x-ray changes of the lung as long as exposure persists, but there is no distinctive fibrosis, no evidence of disability, and no special complicating factors.

# **SECTION 12: Ecological information**

#### **Toxicity**

No data available.

# Persistence and degradability

No data available.

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#### **Bioaccumulative potential**

No data available.

# Mobility in soil

No data available.

#### Results of PBT and vPvB assessment

No data available.

# **Endocrine disrupting properties**

No data available.

#### Other adverse effects

No data available.

# **SECTION 13: Disposal considerations**

# **Disposal methods**

#### Product disposal

Waste material must be disposed of in accordance with the national and local regulations. Leave chemicals in original containers.

## Other disposal recommendations

Do not discharge this material into waterways, drains and sewers.

# **SECTION 14: Transport information**

# ADG (Road and Rail)

Not dangerous goods

# IMDG

Not dangerous goods

# IATA

Not dangerous goods

# **SECTION 15: Regulatory information**

# Safety, health and environmental regulations specific for the product in question

#### **Australia SUSMP**

Poison Schedule: NS

# **New Jersey Right To Know Components**

Common name: TIN CAS number: 7440-31-5

## **Pennsylvania Right To Know Components**

Chemical name: Tin CAS number: 7440-31-5

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**Canadian Domestic Substances List (DSL)** 

Chemical name: Tin CAS: 7440-31-5

## **SECTION 16: Other information**

## Further information/disclaimer

ChemSupply Australia Pty Ltd does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon ChemSupply Australia Pty Ltd with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of ChemSupply Australia Pty Ltd is limited to the replacement of supply of equivalent goods or payment of the cost of replacing the goods or acquiring equivalent goods.