

infosafe CS: 1.7.2

Page: 1 of 6 chem-supply

RE-ISSUED by CHEMSUPP Infosafe No™ 3CH51 Issue Date: July 2017

Product Name: **CESIUM NITRATE**

Classified as hazardous

1. Identification

GHS Product

CESIUM NITRATE

Identifier

CHEM-SUPPLY PTY LTD (ABN 19 008 264 211) **Company Name**

38 - 50 Bedford Street GILLMAN **Address**

SA 5013 Australia

Telephone/Fax Number

Tel: (08) 8440-2000 Fax: (08) 8440-2001

Recommended use of the chemical and restrictions on use

Synthesis of cesium nitratocuprate. Cs2[Cu(NO3)4], which crystallizes in a square-planar coordination around the Cu centre; preparation of other cesium salts; prisms for infrared spectroscopy; infrared flares;

CL541

in x-ray fluorescent screens, scintillation counters; and laboratory reagent.

Other Names Product Code

CESIUM NITRATE LR

Caesium nitrate

Other Information

EMERGENCY CONTACT NUMBER: +61 08 8440 2000 Business hours: 8:30am to 5:00pm, Monday to Friday.

Chem-Supply Pty Ltd does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon Chem-Supply Pty Ltd with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of Chem-Supply Pty Ltd is limited to the replacement of supply of equivalent goods or payment of the cost of replacing the goods or acquiring equivalent goods.

2. Hazard Identification

GHS classification

of the

Oxidizing Solids: Category 1 Acute Toxicity - Oral: Category 4

substance/mixture

Signal Word (s)

Hazard Statement

H271 May cause fire or explosion; strong oxidiser.

H302 Harmful if swallowed.

Pictogram (s) Flame over circle, Exclamation mark,

DANGER





Precautionary statement -

P210 Keep away from heat/sparks/open flames/hot surfaces. - No smoking. P220 Store away from strong reducing agents, easily oxidized materials, alcohols, aluminium and

combustible materials. Prevention

P221 Take any precaution to avoid mixing with combustibles.

P264 Wash thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P283 Wear fire/flame resistant/retardant clothing.

Precautionary statement -Response

P370+P378 In case of fire: Use flooding quantities of water for extinction.

P306+P360 IF ON CLOTHING: rinse immediately contaminated clothing and skin with plenty of water before removing clothes.

P370+P380+P375 In case of fire: Evacuate area. Fight fire remotely due to the risk of explosion. P301+P312 IF SWALLOWED: Call a POISON CENTER or doctor/physician if you feel unwell.

P330 Rinse mouth.

Precautionary statement -**Disposal**

P501 Dispose of contents/container according to local, state and federal regulations.



infosafe CS: 1.7.2

Page: 2 of 6 chem-supply

RE-ISSUED by CHEMSUPP Infosafe No™ 3CH51 Issue Date: July 2017

CESIUM NITRATE Product Name:

Classified as hazardous

3. Composition/information on ingredients

Chemical Solid

Characterization

Ingredients Name CAS **Proportion Hazard Symbol Risk Phrase**

> Cesium nitrate 7789-18-6 100 %

4. First-aid measures

Inhalation Remove from exposure, rest and keep warm. If breathing has stopped, apply artificial respiration. If

breathing is difficult, give oxygen. Seek medical attention.

Rinse mouth thoroughly with water immediately, repeat until all traces of product have been removed. Ingestion

DO NOT INDUCE VOMITING. Seek medical advice if effects persist.

Skin Wash affected area thoroughly with copious amounts of running water. Remove contaminated clothing

and wash before reuse. Seek medical attention.

Eye contact Immediately irrigate with copious quantity of water for at least 15 minutes. Eyelids to be held open.

Seek medical attention.

First Aid Facilities Maintain eyewash fountain and drench facilities in work area.

Treat symptomatically based on judgement of doctor and individual reactions of the patient. **Advice to Doctor**

For advice, contact a Poisons Information Centre (Phone eg Australia 13 1126; New Zealand 0800 764 Other Information

766) or a doctor.

5. Fire-fighting measures

Hazards from Combustion **Products**

Specific Methods

Irritating and highly toxic gases, including oxides of nitrogen and cesium oxide fumes.

Small fire: USE FLOODING QUANTITIES OF WATER. Do not use dry chemicals, CO2 or foam. If safe to do so, move undamaged containers from fire area. Do not move cargo if cargo has been exposed to

Large fire: Flood fire area with water from a protected position. Cool containers with flooding quantities

of water until well after fire is out - If impossible, withdraw from area and let fire burn. Avoid getting water inside containers: a violent reaction may occur. Dam fire control water for later disposal.

Specific hazards arising from the chemical

Will accelerate burning when involved in a fire. May explode from heating, shock, friction or

contamination. May react explosively with hydrocarbons (fuels). May ignite combustibles (wood, paper, clothing, etc). Fire may produce irritating, poisonous, and/or corrosive gases. Containers may explode

when heated. Runoff may create fire or explosion hazard.

Hazchem Code

Decomposition

Decomposes at high temperatures.

Temp.

Precautions in

Wear SCBA and chemical splash suit. Structural firefighter's uniform will provide limited protection.

connection with Fire

6. Accidental release measures

Spills & Disposal Do not contaminate. Keep combustibles (wood, paper, clothing, oil, etc.) away from spilled material. Do

not touch damaged containers or spilled material unless wearing appropriate protective clothing. Use water spray to knock down vapours or divert vapour clouds. Prevent entry into waterways, drains or

confined areas. Prevent exposure to heat.

Dry Spill

Use clean non-sparking tools to transfer material to a clean, dry plastic container and cover loosely.

Move container from spill area.

Small Liquid Spill

Use a non-combustible material like vermiculite, sand or earth to soak up the product and place in a

loosely-covered container for later disposal.

Large Liquid Spill

SEĚK EXPERT ADVICE ON HANDLING AND DISPOSAL.

Use personal protective equipment. Use in a well ventilated area. Avoid dust formation. Avoid breathing **Personal**

dusts, mist, vapours or gas. Evacuate personnel to safe areas. **Precautions**

Personal Protection Wear protective clothing specified for normal operations (see Section 8)

7. Handling and storage



infosafe CS: 1.7.2

Page: 3 of 6 chem-supply

RE-ISSUED by CHEMSUPP Infosafe No™ 3CH51 Issue Date: July 2017

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Classified as hazardous

Handling

Precautions for Safe Avoid ingestion or inhalation of dust. Avoid contact with eyes, skin, and clothing. Avoid prolonged or repeated exposure. Minimize dust generation and accumulation. Use only in a chemical fume hood. Ensure good ventilation at the workplace. Use only in a well-ventilated area. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Wear suitable protective clothing. Remove all soiled and contaminated clothing immediately. Wash thoroughly after handling. Keep away from foodstuffs, beverages and feed. Keep away from incompatibles such as reducing agents, flammable, organic or combustible material. Keep away from heat and all sources of ignition. Ground all equipment containing material. Empty containers pose a fire risk, evaporate the residue under a fume hood.

Conditions for safe storage, including

any incompatabilities Store in suitable, labelled, tightly closed containers, in a cool, dry, well-ventilated area away from incompatible substances. Hygroscopic. Keep well closed and protected from direct sunlight and moisture. Separate from acids, alkalies, reducing agents, flammable substances and combustibles materials. Store away from heat, and from sources of ignition (sparks, and open flame). Avoid any dust build-up by frequent cleaning and suitable construction of storage area. Keep storage separated from work areas. Inspect periodically for deficiencies such as damage or leaks.

Storage Regulations Refer Australian Standard AS 4326-1995 'The storage and handling of oxidizing agents'.

Storage **Temperatures** Store at room temperature (15 to 25 °C recommended).

8. Exposure controls/personal protection

Other Exposure Information

A time weighted average (TWA) concentration for an 8 hour day, and 5 day week has not been established by Safe Work Australia for this product. There is a blanket limit of 10 mg/m3 for dusts when limits have not otherwise been established.

Appropriate

In industrial situations maintain the concentrations values below the TWA. This may be achieved by engineering controls process modification, use of local exhaust ventilation, capturing substances at the source, or other

Respiratory Protection

Where ventilation is not adequate, respiratory protection may be required. Avoid breathing dust, vapours or mists. Respiratory protection should comply with AS 1716 - Respiratory Protective Devices and be selected in accordance with AS 1715 - Selection, Use and Maintenance of Respiratory Protective Devices. Filter capacity and respirator type depends on exposure levels. In event of emergency or planned entry into unknown concentrations a positive pressure, full-facepiece SCBA should be used. If respiratory protection is required, institute a complete respiratory protection program including selection,

Eye Protection Hand Protection fit testing, training, maintenance and inspection. The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate. Must comply with Australian Standards AS 1337 and be selected and used in accordance with AS 1336. Hand protection should comply with AS 2161, Occupational protective gloves - Selection, use and

Personal Protective Equipment Body Protection

Final choice of personal protective equipment will depend on individual circumstances and/or according to risk assessments undertaken.

Clean clothing or protective clothing should be worn, preferably with an apron. Clothing for protection against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals. Always wash hands before smoking, eating or using the toilet. Wash contaminated clothing and other

Hygiene Measures

protective equipment before storing or re-using.

9. Physical and chemical properties

maintenance.

Form

Colourless to off-white, or white crystalline powder or crystals. **Appearance**

Odourless. Odour

Decomposition **Temperature**

Decomposes at high temperatures.

Melting Point 414 °C

Solubility in Water Soluble (150 g/l at 20 °C).

Solubility in Organic Soluble in acetone; slightly soluble in alcohol.

3.68 **Specific Gravity**

Solvents

Flammability May be combustible at high temperature. Strong oxidizer and its heat of reaction with reducing agents

or combustibles may cause ignition. Fire-promoting.



infosafe CS: 1.7.2

Page: 4 of 6 chem-supply

RE-ISSUED by CHEMSUPP Infosafe No™ 3CH51 Issue Date: July 2017

CESIUM NITRATE Product Name:

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Inorganic nitrates act as oxygen carriers which may cause a fire or explosion if in contact with reducing **Explosion**

agents or inorganic matter. **Properties**

A mixture of aluminium powder + water + metal nitrate may explode.

Mixtures of metal nitrates with alkyl exters may explode.

Mixtures of a nitrate with phosphorous, Tin (II) chloride or other reducing agents may react explosively.

Molecular Weight 194.91

Oxidising Properties Strong oxidiser.

10. Stability and reactivity

Stable under normal conditions, temperatures and pressures, but hygroscopic. Oxidizer. Fire-promoting. **Chemical Stability**

Moisture/water, high temperatures, ignition sources, dust generation, combustible materials, organic **Conditions to Avoid**

materials, reducing agents and incompatible materials.

Incompatible Strong reducing agents, easily oxidized materials, alcohols, aluminium, alkyl esters, phosphinates + heat, nitrites, nitrates, organic matter, cyanides (e.g. potassium cyanide, sodium cyanide), thiocyanates, **Materials**

isothiocvanates, hypophosphites, moisture and combustible materials.

Hazardous Irritating and highly toxic gases, including oxides of nitrogen and cesium oxide fumes.

Decomposition **Products**

Possibility of Reacts with reducing agents.

hazardous reactions Reacts with flammable substances. Reactive with reducing agents, organic materials.

Reaction with aluminium powder + water may be explosive.

Reacts with acids.

Hazardous

Reproductive

Will not occur. **Polymerization**

11. Toxicological Information

Ingestion May be harmful if swallowed. May cause digestive tract irritation with nausea, vomiting, abdominal pain,

and diarrhoea (possibly with blood). May cause gastrointestinal tract burns. Large amounts taken by mouth may have serious or even fatal effects. May affect the urinary system (proteinuria), blood (methaemoglobinaemia due to insufficient amount of oxygen in blood). Methaemoglobinaemia may affect behaviour (hyper-irritability, convulsions, spasms, seizures, general depression, headache, mental impairment, dizziness, weakness, fatigue, unconciousness), respiration (dyspnoea), cardiovascular system (blood pressure elevation or lowering, tachycardia or bradycardia, abnormal ECG), and may

cause cvanosis.

May be harmful if inhaled. May cause respiratory tract irritation and respiratory tract problems such as Inhalation

shortness of breath, dyspnoea, pulmonary oedema, asphyxia, chemical pneumonitis, and upper airway

obstruction caused by oedema. May cause blood effects.

May cause mechanical or mild to severe skin irritation and possible burns, resulting in redness and Skin

itching. May be harmful if absorbed through the skin.

Eye May cause eye irritation. May cause conjunctivitis. May cause permanent corneal opacification.

Carcinogenicity Nitrate or nitrite (ingested) under conditions that result in endogenous nitrosation is evaluated in the

IARC Monographs (Vol. 94; in preparation) as Group 2A: Probably carcinogenic to humans.

oxygen debt, animal-testicular **Toxicity**

Nitrates, nitrites & organic nitro compounds [resp/skin/oral]: human-reduced oxygen uptake causing

toxin, abortifacient (From: ôReproductive Hazards of the Workplace" by Linda M. Frazier, MD, MPH &

Marvin L. Hage, MD).

Chronic Effects Prolonged exposure may result in skin burns and ulcerations. Over-exposure by inhalation may cause

respiratory irritation. Repeated or prolonged exposure can cause damage to central nervous system (CNS) and can have blood and neuromuscular effects. Repeated ingestion of product may lead to weakness, general depression, headache and mental impairment. Chronic ingestion of cesium has been

fatal to laboratory animals, possibly due to the replacement of potassium.

Mutagenicity Evidence of mutagenic effects.

12. Ecological information

Ecological No ecological problems are to be expected when the product is handled and used with due care and

Information

Ecotoxicity The following applies to nitrates in general: may contribute to the eutrophication of water supplies.

Hazard for drinking water.



infosafe CS: 1.7.2

Page: 5 of 6 chem-supply

RE-ISSUED by CHEMSUPP Infosafe No™ 3CH51 Issue Date: July 2017

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Environmental

Do not allow to enter waters, waste water, or soil!

Protection

Information

13. Disposal considerations

Dispose of according to relevant local, state and federal government regulations, Disposal

Considerations

14. Transport information

Dangerous Goods of Class 5.1 Oxidising Agents are incompatible in a placard load with any of the **Transport**

following: - Class 1, Class 2.1, Class 2.3, Class 3, Class 4, Class 5.2, Class 7, Class 8, Fire risk

substances and combustible liquids. 1451

U.N. Number

UN proper shipping CAESIUM NITRATE

name

Transport hazard

class(es)

5.1

Hazchem Code

1[Z] 3.8.5.1 **Packaging Method Packing Group** Ш

Storage and **Transport**

Not usually shipped in large quantities.

EPG Number 5A1 **IERG Number** 31

15. Regulatory information

Regulatory

Not listed under WHS Regulation 2011, Schedule 10 - Prohibited carcinogens, restricted carcinogens

and restricted hazardous chemicals. Information

Not Scheduled **Poisons Schedule** Irritant, Oxidising **Hazard Category**

16. Other Information

Literature References

Standard for the Uniform Scheduling of Medicines and Poisons No. 15', Commonwealth of Australia, November 2016.

Lewis, Richard J. Sr. 'Hawley's Condensed Chemical Dictionary 13th. Ed.', Rev., John Wiley and Sons,

National Road Transport Commission, 'Australian Code for the Transport of Dangerous Goods by Road and Rail 7th. Ed.', 2007.

Safe Work Australia, 'National Code of Practice fot the Preparation of Safety Data Sheets for Hazardous Chemicals', 2011.

Standards Australia, 'SAA/SNZ HB 76:2010 Dangerous Goods - Initial Emergency Response Guide', Standards Australia/Standards New Zealand, 2010.

Safe Work Australia, 'Approved Criteria for Classifying Hazardous Substances [NOHSC:1008 (2004)]'. Safe Work Australia, 'Hazardous Substances Information System, 2005'.

Safe Work Australia, 'National Code of Practice for the Labelling of Safe Work Hazardous Substances

Safe Work Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational

Environment [NOHSC:1003(1995) 3rd Edition]'. Paul McCarthy Ph. (08) 8440 2000 DISCLAIMER STATEMENT:

Contact Person/Point

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Empirical Formula & Structural Formula

CsNO3

...End Of MSDS...





chem-supply Page: 6 of 6

Infosafe No™ 3CH51 Issue Date :July 2017 RE-ISSUED by CHEMSUPP

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