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Infosafe No™ 1CHCC Issue Date : February 2022 RE-ISSUED by CHEMSUPP

Product Name COBALT (II) SULFATE Heptahydrate

Classified as hazardous

#### **Section 1 - Identification**

COBALT (II) SULFATE Heptahydrate **Product Identifier** 

CHEMSUPPLY AUSTRALIA PTY LTD (ABN 19 008 264 211) **Company Name** 

38 - 50 Bedford Street GILLMAN Address

SA 5013 Australia Tel: (08) 8440-2000 Telephone/Fax

Number

**Emergency Phone** 

CHEMCALL 1800 127 406 (Australia) / +64-4-917-9888 (International)

Number

E-mail Address www.chemsupply.com.au

the chemical and restrictions on use

Recommended use of Ceramics, pigments, glazes, in plating baths for cobalt, agricultural industry, animal feed additive, additive to soils, catalyst, paint and ink drier, storage batteries, analytical reagent and laboratory reagent.

Other Names Name Product Code

> COBALT (II) SULFATE Heptahydrate AR CA085 COBALT (II) SULFATE Heptahydrate LR CL085

Cobalt sulfate, Cobaltous sulfate,

Bieberite

CT649 COBALT(II) SULFATE Monohydrate

#### Other Information

ChemSupply Australia Pty Ltd does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon ChemSupply Australia Pty Ltd with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of ChemSupply Australia Pty Ltd is limited to the replacement of supply of equivalent goods or payment of the cost of replacing the goods or acquiring equivalent goods.

#### Section 2 - Hazard(s) Identification

**GHS Classification** 

of the

 ${\tt Hazardous}$  to the Aquatic Environment - Acute Hazard: Category 1 Hazardous to the Aquatic Environment - Long-Term Hazard: Category 1 Carcinogenicity: Category 1

Substance/Mixture

Acute Toxicity - Oral: Category 4 Sensitization - Skin: Category 1 Toxic to Reproduction: Category 1 Sensitization - Respiratory: Category 1 Germ Cell Mutagenicity: Category 2

DANGER Signal Word

**Hazard Statement (s)** 

H302 Harmful if swallowed.

H317 May cause an allergic skin reaction.

H334 May cause allergy or asthma symptoms or breathing difficulties if

inhaled.

H341 Suspected of causing genetic defects.

H350 May cause cancer.

H360 May damage fertility or the unborn child.

H410 Very toxic to aquatic life with long lasting effects.

Health hazard, Exclamation mark, Environment Pictogram (s)











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P201 Obtain special instructions before use. **Precautionary** P202 Do not handle until all safety precautions have been read and understood. Statement -P261 Avoid breathing dust/fume/gas/mist/vapours/spray. Prevention P264 Wash thoroughly after handling. P270 Do not eat, drink or smoke when using this product. P272 Contaminated work clothing should not be allowed out of the workplace. P273 Avoid release to the environment. P280 Wear protective gloves/protective clothing/eye protection/face protection. P284 Wear respiratory protection. P301+P312 IF SWALLOWED: Call a POISON CENTER or doctor/physician if you feel **Precautionary** unwell. Statement -P330 Rinse mouth. Response P302+P352 IF ON SKIN: Wash with plenty of soap and water. P363 Wash contaminated clothing before reuse. P333+P313 If skin irritation or rash occurs: Get medical advice/attention. P304+P341 IF INHALED: If breathing is difficult, remove victim to fresh air and keep at rest in a position comfortable for breathing. P342+P311 If experiencing respiratory symptoms: Call a POISON CENTER or

doctor/physician. P308+P313 IF exposed or concerned: Get medical advice/attention.

P405 Store locked up. **Precautionary** 

Statement - Storage

**Precautionary** Statement - Disposal P501 Dispose of contents/container to an approved waste disposal plant.

#### **Section 3 - Composition and Information on Ingredients**

Ingredients	Name	CAS	Proportion
	Cobalt (II) sulfate heptahydrate	10026-24-1	96-100 %

### **Section 4 - First Aid Measures**

Inhalation	If inhaled, remove from contaminated area to fresh air immediately. Apply artificial respiration if not breathing. If breathing is difficult, give oxygen. Consult a physician.		
Ingestion	Rinse mouth thoroughly with water immediately, repeat until all traces of product have been removed. DO NOT INDUCE VOMITING. Seek immediate medical advice.		
Skin	Wash with plenty of soap and water. Remove contaminated clothing and wash before re-use. Seek medical attention.		
Eye	Immediately irrigate with copious quantity of water for at least 15 minutes. Eyelids to be held open. Seek medical attention.		
First Aid Facilities	Maintain eyewash fountain and drench facilities in work area.		
Advice to Doctor	Treat symptomatically based on judgement of doctor and individual reactions of		

the patient.

Most important Allergic reactions, diarrhoea.

Symptoms of an acute cobalt intoxication: diarrhoea, loss of appetite, drop in symptoms/effects, body temperature, drop in blood pressure. Toxic effect on kidneys acute, delayed and (proteinuria, anuria), heart and pancreas. aggravated medical

conditions

**Other Information** 

For advice, contact a Poisons Information Centre (Phone eg Australia 13 1126; New Zealand 0800 764 766) or a doctor.

## **Section 5 - Firefighting Measures**

Toxic fumes of sulfur oxides (SO2, SO3, SOx) and cobalt oxides. Hazards from Combustion **Products** 

**Specific Methods** Use extinguishing media most appropriate for the surrounding fire. No

limitations to the type of extinguishing media.

Small fire: Use dry chemical, CO2, water spray or foam.





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Large fire: Use water spray, fog or foam.

If safe to do so, move undamaged containers from the fire area. Cool

containers with flooding quantities of water until well after the fire is out.

**Hazchem Code** 

**Decomposition Temperature** 

At 41.5 °C the substance dehydrates from heptahydrate to hexahydrate, at 71 °C to monohydrate; above 600 °C decomposition occurs to cobalt oxide (CoS04), this decomposes to CoO at 900-950 °C. Oxides of sulphur will be generated on this decomposes to CoO at 900-950 heating above 600 °C.

**Precautions in** connection with Fire

**Other Information** 

Wear SCBA and structural firefighter's uniform.

Do NOT use halogenated type extinguishers as catalytic decomposition of the

extinguishing medium may well occur with the production of toxic gases, including phosgene.

Suppress gases/vapours/mist with a water spray jet. Prevent fire extinguishing

water from contaminating surface water or ground water.

#### **Section 6 - Accidental Release Measures**

Avoid substance contact. Avoid generation of dusts: do not inhale dusts. **Personal Precautions** 

Ensure supply of fresh air in enclosed rooms.

Wear protective clothing specified for normal operations (see Section 8) **Personal Protection** 

Clean-up Methods -**Small Spillages** 

Sweep up (avoid generating dust) and using clean non-sparking tools transfer to a clean, suitable, clearly labelled container for disposal in accordance with local regulations.

Clean-up Methods -Large Spillages

Stop leak if safe to do so. Prevent entry into waterways, drains, confined areas. Prevent dust cloud. Use clean non-sparking tools to collect material and place it into loosely-covered plastic containers for later disposal.

## Section 7 - Handling and Storage

**Precautions for Safe** Handling

Avoid ingestion and inhalation. Work under a fume hood. Do not breathe dust. Avoid contact with eyes, skin, or clothing. Avoid prolonged or repeated exposure. Minimize dust generation and accumulation. Keep container tightly closed. Use only with adequate ventilation. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Wear suitable protective clothing. Under no circumstances eat, drink or smoke while handling this material. Wash thoroughly after handling. Contaminated clothing should be removed and washed before re-use.

Conditions for safe storage, including any incompatibilities Store in tightly closed containers, in a cool, dry, well-ventilated area away from incompatible substances. Separated from strong oxidants. Protect against physical damage, direct sunlight and moisture. Keep away from heat. Containers of this material may be hazardous when empty since they retain product residues (dust, solids); observe all warnings and precautions listed for the product.

Storage

Store at room temperature (15 to 25 °C recommended).

**Temperatures** 

#### **Section 8 - Exposure Controls and Personal Protection**

TWA STEL Occupational Name **Exposure Limit** (OEL) Values

> mg/m3mg/m3 Footnote ppm ppm 0.05 Cobalt (II) sulfate Cobalt, heptahydrate metal dust & fume (as

Other Exposure Information

A time weighted average (TWA) has been established for Cobalt, metal dust & fume (as Co) (Safe Work Australia) of 0.05 mg/m3. The exposure value at the TWA is the average airborne concentration of a particular substance when calculated over a normal 8 hour working day for a 5 day working week. Note: Sensitiser.

TWA - the Time-Weighted Average airborne concentration over an eight-hour



Respiratory

**Protection** 

## Material Safety Data Sheet



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working day, for a five-day working week over an entire working life.

'Sen' notice - sensitiser. The substance can cause a specific immune response in some people. An affected individual may subsequently react to minute levels

of that substance.

In industrial situations maintain the concentrations values below the TWA. **Engineering** 

This may be achieved by process modification, use of local exhaust

**Controls** ventilation, capturing substances at the source, or other methods.

Where ventilation is not adequate, respiratory protection may be required. Avoid breathing dust, vapours or mists. Respiratory protection should comply with AS 1716 - Respiratory Protective Devices and be selected in accordance with AS 1715 - Selection, Use and Maintenance of Respiratory Protective Devices. Filter capacity and respirator type depends on exposure levels. event of emergency or planned entry into unknown concentrations a positive pressure, full-facepiece SCBA should be used. If respiratory protection is

required, institute a complete respiratory protection program including selection, fit testing, training, maintenance and inspection.

The use of a face shield, chemical goggles or safety glasses with side shield **Eye and Face** protection as appropriate. Must comply with Australian Standards AS 1337 and **Protection** 

be selected and used in accordance with AS 1336.

Hand protection should comply with AS 2161, Occupational protective gloves -**Hand Protection** 

Selection, use and maintenance. Nitrile rubber gloves

Final choice of personal protective equipment will depend on individual **Personal Protective** 

circumstances and/or according to risk assessments undertaken. **Equipment** 

Safety boots in industrial situations is advisory, foot protection should **Footwear** 

comply with AS 2210, Occupational protective footwear - Guide to selection, care and use.

Clean clothing or protective clothing should be worn, preferably with an **Body Protection** 

apron. Clothing for protection against chemicals should comply with AS 3765

Clothing for Protection Against Hazardous Chemicals.

Always wash hands before smoking, eating or using the toilet. Wash **Hygiene Measures** 

contaminated clothing and other protective equipment before storing or

re-using.

Section 9 - Physical and Chemical Properties

Solid **Form** 

**Appearance** Rose-coloured, monoclinic crystals (monohydrate).

Pink to red monoclinic, pristmatic crystals (heptahydrate).

Odourless. Odour

Loses water molecules at 41.5 °C, 71 °C and above 600 °C. CoS04 decomposes to **Melting Point** 

CoO at 900-950  $^{\circ}\text{C}$  and melts at ~1800  $^{\circ}\text{C}$ .

At 41.5  $^{\circ}\text{C}$  the substance dehydrates from heptahydrate to hexahydrate, at 71  $^{\circ}\text{C}$ **Decomposition** to monohydrate; above 600 °C decomposition occurs to cobalt oxide (CoS04), this decomposes to CoO at 900-950 °C. Oxides of sulphur will be generated on heating above 600 °C. **Temperature** 

Soluble (362 g/1 at 20°C). Solubility in Water

**Solubility in Organic** Slightly soluble in methanol, ethanol (heptahydrate).

Solvents

**Specific Gravity** 1.948

~4 (100 g/l, H2O, 20 °C) pН

Negligible. Vapour Pressure 0 %vol @ 21 °C **Volatile Component** 

Non combustible material. **Flammability** 

281.10 Molecular Weight

REFRACTIVE INDEX: 1.483 **Other Information** 

Section 10 - Stability and Reactivity





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Stable under ordinary conditions of use and storage. Releases one water of **Chemical Stability** 

crystallisation when heated to 41.5 °C.

Dust generation, moisture and incompatible materials. **Conditions to Avoid** 

Incompatible Materials

Strong oxidizing agents, tert-butyl hydroperoxide.

Hazardous Decomposition

Toxic fumes of sulfur oxides (SO2, SO3, SOx) (possibly sulfuric acid) and

cobalt oxides.

**Products** Hazardous

Will not occur.

**Polymerization** 

#### Section 11 - Toxicological Information

Acute Toxicity - Oral LD50 (rat): 582 mg/kg (heptahydrate).

Toxic if swallowed. May cause gastrointestinal irritation with abdominal pain, Ingestion

nausea, vomiting, diarrhoea, flushing of the face and ears, mild hypotension, rash, and ringing in the ears. May have cumulative toxic action where elimination cannot keep pace with absorption. Large amounts depress erythrocyte production and may have adverse effects on the thyroid. Toxic effect on the kidneys, heart and pancreas. Symptoms of acute cobalt intoxication include lack of appetite, drop in blood pressure, agitation, spasms. A single case of poisoning, liver and kidney damage has been

attributed to cobalt. There have also been reports of haemotologic, digestive and pulmonary changes in humans.

Harmful if inhaled. Inhalation of dust may cause irritation to nose, throat Inhalation

and the respiratory tract. Symptoms may include coughing, sore throat, laboured breathing, shortness of breath and nausea. Risk of airways

sensitisation. Respiratory hypersensitivity, asthma may appear. Inhalation of dust may cause cancer (animal data). Inhalation of cobalt dust and fumes is

associated with an increased incidence of lung disease.

Causes irritation to skin. Symptoms include redness, itching, and pain. May Skin cause dermatitis. Risk of sensitisation, an allergic skin reaction, which

becomes evident upon re-exposure to this material. May be harmful if absorbed

through the skin.

Causes irritation, redness, and pain. Eve

Sensitisation: Sensitisation test (guinea pig): Positive. **Skin Sensitisation** 

Cobalt sulfate [10026-24-1] and other soluble cobalt(II) salts are evaluated Carcinogenicity

in the IARC Monographs (Vol. 86; 2006) as Group 2B: Possibly carcinogenic to

humans.

Group 2B: The agent is possibly carcinogenic to humans.

This category is used for agents for which there is limited evidence of carcinogenicity in humans and less than sufficient evidence of carcinogenicity in experimental animals. It may also be used when there is inadequate evidence

of carcinogenicity in humans but there is sufficient evidence of

carcinogenicity in experimental animals. In some instances, an agent for which there is inadequate evidence of carcinogenicity in humans and less than sufficient evidence of carcinogenicity in experimental animals together with supporting evidence from mechanistic and other relevant data may be placed in this group. An agent may be classified in this category solely on the basis of

strong evidence from mechanistic and other relevant data.

Repeated oral administration may produce goiter and reduced thyroid activity. **Chronic Effects** 

Prolonged or repeated skin exposure may cause dermatitis. Repeated or prolonged contact may cause skin sensitization. Repeated or prolonged inhalation exposure may cause asthma. Repeated or prolonged exposure to the substance can produce damage to the respiratory tract, lungs, kidneys, heart

and bone marrow, resulting in cardiomyopathy, pericardial effusion,

polycardial effusion, polycythemia, cardiac failure, vomiting and convulsions. This substance is possibly carcinogenic to humans. Animal tests show that this substance possibly causes toxic effects upon human reproduction. Animal tests show that this substance possibly causes malformations in human babies. May

damage fertility.





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### Section 12 - Ecological Information

Highly toxic for aquatic organisms. May cause long-term adverse effects in **Ecotoxicity** 

the aquatic environment.

Persistence and **Degradability** 

Methods for the determination of biodegradability are not applicable to

inorganic substances.

**Environmental** 

**Protection** 

Do not allow to enter waters, waste water, or soil!

## **Section 13 - Disposal Considerations**

**Disposal** Considerations Whatever cannot be saved for recovery or recycling should be handled as hazardous waste and disposed of according to relevant local, state and federal government regulations.

## **Section 14 - Transport Information**

**Transport** Information

Class 9 Miscellaneous dangerous goods shall not be loaded in a vehicle with: Class 1 Explosives - Class 5. 1 Oxidizing agents (when Class 9 substance capable of igniting and burning - Class 5. 2 Organic peroxides (when Cl. capable of igniting/burnin Environmentally Hazardous Substances meeting the descriptions of UN 3077 or UN 3082 are not subject to this Code when

transported by road or rail in;

(a) packagings that do not incorporate a receptacle exceeding 500 kg(L); or

(b) IBCs.

ADG UN Number

3077

**ADG Proper Shipping Name**  ENVIRONMENTALLY HAZARDOUS SUBSTANCE, SOLID, N.O.S. - (Cobalt (II) Sulfate

Heptahydrate)

**ADG Transport Hazard Class** 

**ADG Packing Group** III **Hazchem Code** 2Z**EPG Number** 9C1 IERG Number

Environmental Hazards

Highly toxic to aquatic organisms. May cause long term adverse effects in the

aquatic environment.

#### **Section 15 - Regulatory Information**

Regulatory

Listed in the Australian Inventory of Chemical Substances (AICS).

Information

**Poisons Schedule** 

Not Scheduled

### Section 16 - Any Other Relevant Information

Literature References

'Standard for the Uniform Scheduling of Medicines and Poisons .', Commonwealth of Australia.

National Road Transport Commission, 'Australian Code for the Transport of Dangerous Goods by Road and Rail 7th. Ed.'.

Safe Work Australia, 'National Code of Practice for the Preparation of Safety Data Sheets for Hazardous Chemicals'

Standards Australia, 'SAA/SNZ HB 76:2010 Dangerous Goods - Initial Emergency Response Guide', Standards Australia/Standards New Zealand.

Safe Work Australia, 'Hazardous Chemical Information System'

Safe Work Australia, 'National Code of Practice for the Labelling of Safe Work Hazardous Substances'.

Safe Work Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational Environment'.

Contact Person/Point Paul McCarthy Ph. (08) 8440 2000 DISCLAIMER STATEMENT:

All information provided in this data sheet or by our technical

representatives is compiled from the best knowledge available to us. However, since data, safety standards and government regulations are subject to change





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Empirical Formula & Structural Formula

CoSO4.7H2O

...End Of MSDS...

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