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RE-ISSUED by CHEMSUPP Infosafe No™ 1CHHL Issue Date: December 2019

COPPER METAL Powder Product Name:

Classified as hazardous

1. Identification

GHS Product

COPPER METAL Powder

Identifier

CHEM-SUPPLY PTY LTD (ABN 19 008 264 211) **Company Name**

38 - 50 Bedford Street GILLMAN **Address**

SA 5013 Australia

Telephone/Fax Number

Tel: (08) 8440-2000 Fax: (08) 8440-2001

Emergency phone

CHEMCALL 1800 127 406 (Australia) / +64-4-917-9888 (International)

number

Recommended use of the chemical and restrictions on use Other Names

Electrical conductors, ammunition, manufacture of bronzes, brass and other copper alloys, electroplated protective coatings and undercoats for nickel, chromium and zinc, insecticides, catalyst, antifouling

paints and laboratory reagent.

<u>Name</u> **Product Code** CL051

COPPER METAL Powder Other Information

> Chem-Supply Pty Ltd does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon Chem-Supply Pty Ltd with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of Chem-Supply Pty Ltd is limited to the replacement of supply of equivalent goods

or payment of the cost of replacing the goods or acquiring equivalent goods.

2. Hazard Identification

GHS classification of the

Hazardous to the Aquatic Environment - Acute Hazard: Category 1 Hazardous to the Aquatic Environment - Long-Term Hazard: Category 1

substance/mixture

Flammable Solids: Category 1

DANGER Signal Word (s)

Hazard Statement

(s)

H228 Flammable solid.

H400 Very toxic to aquatic life.

H410 Very toxic to aquatic life with long lasting effects.

Flame, Environment Pictogram (s)





Precautionary

P210 Keep away from heat/sparks/open flames/hot surfaces. - No smoking.

statement -Prevention

P240 Ground/bond container and receiving equipment. P241 Use explosion-proof electrical/ventilating/lighting/.../equipment.

P273 Avoid release to the environment.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

Precautionary statement -

P370+P378 In case of fire: Use dry sand, dry chemical or alcohol-resistant foam for extinction.

P391 Collect spillage.

Response **Precautionary**

P501 Dispose of contents/container to an approved waste disposal plant.

statement -Disposal

3. Composition/information on ingredients

Solid

Chemical

Characterization

Ingredients **Proportion Hazard Symbol** Name CAS

> 7440-50-8 100 % Copper



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4. First-aid measures

Ingestion

Products

Inhalation If inhaled, remove from contaminated area to fresh air immediately. Apply artificial respiration if not

breathing. If breathing is difficult, give oxygen. Get medical aid if cough or other symptoms appear.

Rinse mouth thoroughly with water immediately, repeat until all traces of product have been removed.

DO NOT INDUCE VOMITING. Seek medical advice if effects persist.

Skin Wash affected areas with copious quantities of water. If irritation occurs seek medical advice.

Eve contact Immediately irrigate with copious quantity of water for at least 15 minutes. Eyelids to be held open.

Seek medical advice if effects persist.

First Aid Facilities Eve wash fountains and safety showers should be available for emergency use.

Advice to Doctor Treat symptomatically based on judgement of doctor and individual reactions of the patient.

Other Information For advice, contact the National Poisons Information Centre (Phone Australia 13 11 26; New Zealand

0800 764 766) or a doctor.

5. Fire-fighting measures

Hazards from Irritating, toxic and corrosive fumes and vapours including copper fumes and some metallic oxides. Combustion

Do NOT use halogenated hydrocarbon extinguishers. **Specific Methods**

In case of fire: Use dry sand, dry chemical or alcohol-resistant foam for extinction.

If safe to do so, move undamaged containers from the fire area. Cool containers with flooding quantities

of water until well after the fire is out.

Specific hazards arising from the chemical

May burn but not ignite readily. May be ignited by friction, heat, sparks or flame. Vapours or dust may form explosive mixtures with air. May burn fiercely. May re-ignite after fire is extinguished. Fire may produce irritating, poisonous and/or corrosive gases. Runoff may pollute waterways. Containers may

explode when heating.

Hazchem Code

Precautions in

Wear SCBA and structural firefighter's uniform.

connection with Fire

6. Accidental release measures

1Z

Avoid breathing dust or vapours and contact with skin and eyes. Vacuum or sweep up material and Spills & Disposal

place into a suitable container for disposal. Avoid creating dusty conditions. Provide ventilation.

Follow precautions for safe handling described in this safety data sheet. No smoking, sparks, flames or Personal other sources of ignition near spillage. Provide adequate ventilation. Keep unnecessary and unprotected **Precautions**

personnel away from the spillage. Treat the spilled material according to the instructions in the clean-up

section.

Avoid contact with skin and eyes. Avoid ingestion and or inhalation of material.

Personal Protection Wear protective clothing specified for normal operations (see Section 8)

Clean-up Methods -Small Spillages

Sweep up (avoid generating dust) and remove to a suitable, clearly labelled container for disposal in

accordance with local regulations.

Environmental Prevent from entering into drains, ditches or rivers.

Precautions

7. Handling and storage

Handling

Precautions for Safe Avoid generation or accumulation of dusts. Avoid prolonged or repeated contact with skin, eyes and clothing. Keep container tightly closed when not in use. Use in well ventilated areas away from all

ignition sources. In case of insufficient ventilation, wear suitable respiratory equipment.

Conditions for safe storage, including

Store away from sources of heat or ignition. Store in a cool, dry place. Isolate from incompatible substances.

incompatabilities Corrosiveness

More resistant to atmospheric corrosion than iron, forming a green layer of hydrated basic carbonate.

Readily attacked by alkalies. Attacked by acetic acid and other organic acids.

Storage Regulations Refer Australian Standard AS/NZS 5026-2012 'The storage and handling of Class 4 dangerous goods'.

8. Exposure controls/personal protection



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Name

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Occupational exposure limit values

STEL TWA

 mg/m3
 ppm
 mg/m3
 ppm
 Footnote

 Copper
 0.2
 Copper (fume)

Other Exposure Information

These Workplace Exposure Standards are guides to be used in the control of occupational health hazards. All atmospheric contamination should be kept to as low a level as is workable. These workplace exposure standards should not be used as fine dividing lines between safe and dangerous

concentrations of chemicals. They are not a measure of relative toxicity.

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concentrations of chemicals. They are not a measure of relative toxicity.

The exposure value at the TWA is the average airborne concentration of a particular substance when

calculated over a normal 8 hour working day for a 5 day working week.

Appropriate engineering controls

Maintain the concentrations values below the TWA. This may be achieved by process modification, use

engineering controls of local exhaust ventilation, capturing substances at the source, or other methods.

Respiratory
Protection

Where ventilation is not adequate, respiratory protection may be required. Avoid breathing dust, vapours or mists. Respiratory protection should comply with AS 1716 - Respiratory Protective Devices and be selected in accordance with AS 1715 - Selection, Use and Maintenance of Respiratory Protective Devices. Filter capacity and respirator type depends on exposure levels. In event of emergency or planned entry into unknown concentrations a positive pressure, full-facepiece SCBA should be used. If respiratory protection is required, institute a complete respiratory protection program including selection,

fit testing, training, maintenance and inspection.

Eye Protection The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate.

Must comply with Australian Standards AS 1337 and be selected and used in accordance with AS 1336.

Hand Protection

Wear gloves of impervious material conforming to AS/NZS 2161: Occupational protective gloves Selection, use and maintenance. Final choice of appropriate glove type will vary according to individual

circumstances. This can include methods of handling, and engineering controls as determined by appropriate risk assessments. Avoid skin contact when removing gloves from hands, do not touch the

gloves outer surface. Dispose of gloves as hazardous waste.

Personal Protective Equipment

Personal protective equipment should not solely be relied upon to control risk and should only be used when all other reasonably practicable control measures do not eliminate or sufficiently minimise risk. Guidance in selecting personal protective equipment can be obtained from Australian, Australian/New

Zealand or other approved standards.

Footwear Safety boots in industrial situations is advisory, foot protection should comply with AS 2210,

Occupational protective footwear - Guide to selection, care and use.

Body Protection Clean clothing or protective clothing should be worn, preferably with an apron. Clothing for protection

against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals.

Hygiene Measures Always wash hands before smoking, eating or using the toilet. Wash contaminated clothing and other

protective equipment before storing or re-using.

9. Physical and chemical properties

Form Solid

Appearance Distinctive reddish coloured powder.

OdourOdourless.Melting Point1083 °CBoiling Point2580 - 2595 °CSolubility in WaterInsoluble.

Solubility in Organic Slowly soluble in ammonia water, slightly soluble in diluted acid.

Solvents

Specific Gravity 8.94

Vapour Pressure 1 mm Hg @ 1628 °C

Flammability Combustible.

Molecular Weight 63.55

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Excellent conductor of electricity. Dissolves readily in nitric and hot concentrated sulfuric acids, in Other Information

hydrochloric and dilute sulfuric acids slowly, but only when exposed to the atmosphere.

Heat of fusion: 48.9 cal/g. Heat of vapourisation: 300.3 kJ/mol. Mohs' hardness: 3.0.

Resistivity: 1.673 microohms/cm.

10. Stability and reactivity

Chemical Stability Stable under normal use conditons.

When exposed to air/moist over a long period of time, the copper becomes dull in colour and gradually

becomes coated with green basic carbonate.

Conditions to Avoid Prolonged exposure to air and moisture and incompatibles.

Incompatible **Materials**

Acetylene, air, alkali oxides, ammonium nitrate, bromates, bromopropyne, chlorates, iodates, chlorine, chlorine trifluorine, (chloride + oxygen difluoride), ethylene oxide, fluorine, halogens and halogenated compounds, hydrogen peroxide, hydrogen sulfide, hydrazine mononitrate, hydrazoic acid, hydrogen sulfide, lead azide, nitrates, nitrides, oxidisers, peroxides, phosphorus, picrates, potassium peroxide,

sodium azide, sodium peroxide, sulfur, sulfuric acid.

Hazardous Decomposition **Products**

Irritating, toxic and corrosive fumes and vapours including copper fumes and some metallic oxides.

Possibility of

Reacts violently with acetylene, ammonium nitrate, bromates, bromopropyne, chlorates, iodates, hazardous reactions chlorine, chlorine trifluorine, (chloride + oxygen difluoride), ethylene oxide, fluorine, halogens and halogenated compounds, hydrogen peroxide, hydrazine mononitrate, hydrazoic acid, hydrogen sulfide, lead azide, nitrates, nitrides, peroxides, potassium peroxide, sodium azide and sodium peroxide.

Reaction with strong acids releases flammable hydrogen gas.

Copper ignites on contact with chlorine, fluorine (abouve 121°C), chlorine trifluoride, and hydrazinum

nitrate (above 70°C).

Hazardous Polymerization Will not occur.

11. Toxicological Information

Ingestion

Ingestion of sufficient concentrations may cause irritation and possible burning and pain of mucous membranes in the mouth, pharynx, oesophagus, and gastrointestinal tract. Poisoning could occur due to the material being soluble in hydrochloric acid, which the stomach contains. Symptoms include salvation, headache, nausea, abdominal/gastric pain, diziness, tiredness, metallictaste, convulsions, jaundice, shock, bloodly diarrhoea and vomiting (vomitus is characteristically greenish-blue) may occur. If vomiting does not occur immediately systemic copper poisoning may occur, such as capillary damage, headache, cold sweat, weak pulse, ulceration in the stomach and intestines, internal haemorrhage,

nephritis, coma and possibly even death.

Inhalation Inhalation of dust and fumes of metallic copper causes irritation to the mucous membranes of the upper

> respiratory tract (nose, throat, mouth). Early symptoms of copper poisoning include coughing, sore throat, wheezing, high temperatures, metallic taste, nausea, general weakness, muscle aches and exhaustion) - metal fume fever. May result in harmful corrosive effects including lesions, ulcerations and perforation of the nasal septum and respiratory tract, delayed pulmonary edema, pneumonitis and emphysema. Copper poisoning by inhalation lead to haemolysis of the red blood cells, hepatic necrosis,

gastro intestinal bleeding, ozotemia, hemoglobinuria, coma and death.

May cause discolouration of the skin; greenish-black skin. May be harmful if absorbed through the skin. Skin

> Causes skin irritation, possibly severe, resulting in redness, itching and pain. May cause skin sensitization, an allergic reaction, which becomes evident upon re-exposure to this material.

Copper (dust/mist) particles may cause eye irritation with symptoms including redness, itching, pain, stinging, blurred vision, discoloration and possible eye damage (permanent corneal opacifiaction,

chemical conjunctivitis, ulceration) leading to irreversible eve injury.

Respiratory sensitisation

Eye

Not classified based on available information.

Skin Sensitisation Not classified based on available information. Germ cell Not classified based on available information.

mutagenicity Carcinogenicity

Not classified as a human carcinogen.

Reproductive Not classified based on available information. **Toxicity**



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STOT-single exposure

Not classified based on available information.

STOT-repeated exposure

Not classified based on available information.

Aspiration Hazard

Not classified based on available information.

Chronic Effects

Prolonged or repeated skin exposure may cause defatting leading to dermatitis. Prolonged or repeated exposure to copper (dust/mist) may cause discolouration of the skin or hair, blood and liver damage, ulceration and perforation of the nasal septum, runny nose, metallic taste, and atrophic changes and irritation of the mucous membranes. Effects may be delayed. Individuals with Wilson's disease are unable to metabolize copper. Thus, copper accumulates in various tissues and may result in liver, kidney and brain damage. Chronic copper poisoning is typified by brain damage and demyelination, kidney defects, and copper deposition in the cornea as exemplified by humans with Wilson's disease. It has also been reported that copper 'metal fever' poisoning has lead to hemolytic anemia and accelerates arteriosclerosis. Symptoms of systemic copper poisoning may include: capillary damage, headache, cold sweat, weak pulse, and kidney and liver damage, central nervous system excitation followed by depression, jaundice, convulsions, paralysis, and coma. Death may occur from shock or renal failure. Depending on the intensity and duration of exposure, effects may vary from mild irritation to severe

destruction of tissue.

The lethal dose of an untreated adult is 10 - 20 g Copper.

Mutagenicity No evidence of mutagenic properties.

12. Ecological information

Ecological Information Due to the poor solubility of the product, no harmful effects on plants and/or aquatic organisms are to be

expected when handled and used with due care and attention.

Ecotoxicity Quantitative data on the ecological effect of this product are not available.

Persistence and degradability

Not degradable in water.

13. Disposal considerations

Disposal Considerations

Whatever cannot be saved for recovery or recycling should be disposed of according to relevant local, state and federal government regulations.

14. Transport information

Transport

- Class 4. 1 Flammable solids - Class 4. 2 Spontaneously combustible substances - Class 4. 3

Information

Dangerous when wet substances - Class 5. 2 Organic peroxides - Class 6 Poisonous (toxic) substances

(capable of igniting/burning

U.N. Number

3089

UN proper shipping METAL POWDER, FLAMMABLE, N.O.S. - (COPPER POWDER)

name

Transport hazard

class(es)

4.1

Hazchem Code Packing Group 1Z

Ш **EPG Number** 4A1 **IERG Number** 29

15. Regulatory information

Regulatory Information Listed in the Australian Inventory of Chemical Substances (AICS). Not listed under WHS Regulation 2011. Schedule 10 - Prohibited carcinogens, restricted carcinogens and restricted hazardous chemicals.

Not Scheduled **Poisons Schedule**

16. Other Information

Literature References

Standard for the Uniform Scheduling of Medicines and Poisons .', Commonwealth of Australia. Lewis, Richard J. Sr. 'Hawley's Condensed Chemical Dictionary 13th. Ed.', Rev., John Wiley and Sons,

Inc., NY, 1997.

National Road Transport Commission, 'Australian Code for the Transport of Dangerous Goods by Road and Rail 7th. Ed.', 2007.

Safe Work Australia, 'National Code of Practice fot the Preparation of Safety Data Sheets for Hazardous





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Standards Australia/Standards New Zealand, 2010.

Safe Work Australia, 'Approved Criteria for Classifying Hazardous Substances [NOHSC:1008 (2004)]'.

Safe Work Australia, 'Hazardous Chemical Information System, 2005'.

Safe Work Australia, 'National Code of Practice for the Labelling of Safe Work Hazardous Substances

2011)

Safe Work Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational

Environment [NOHSC:1003(1995) 3rd Edition]'.

Contact Person/Point Paul McCarthy Ph. (08) 8440 2000 DISCLAIMER STATEMENT:

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information provided in this data sheet or by our technical representatives.

Empirical Formula & Cu **Structural Formula**

Other Information December 2014 changed to a dangerous good Class 4.1 Fammable solid.

...End Of MSDS...

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